

Given the following, find the number of grams:

9) 4 moles of $\text{Cu}(\text{CN})_2$

10) 5.6 moles of C_6H_6

11) 21.3 moles of BaCO_3

12) 1.2 moles of $(\text{NH}_4)_3\text{PO}_3$

13) 9.3×10^{-3} moles of SmO

14) 6.6 moles of ZnO

15) 5.4 moles of K_2SO_4

16) 88.4 moles of Ni_3